## Quantitative analysis of ascorbic acid with lodine titrant

## Use

This method is used to determine the content of ascorbic acid (Vitamin C) with the lodine titrant in juices like orange or apple juice. The sulfite $\left(\mathrm{SO}_{2}\right)$ in the juice is masked before with glyoxal solution.

## Appliances

- Titrator: TL 6000/7000 (TL 6000/7000 M1/10) consists of
- Basic device
- Magnetic stirrer TM 235
- 10 mL Exchange unit WA 10, with brown glass bottle for titrant complete


## Electrodes

- Electrode: Pt 1200 or with cable L 1 NN


## Reagents

- Titration agent: Iodine solution $0.01 \mathrm{~mol} / \mathrm{L}$
- Other reagents: $\mathrm{H}_{2} \mathrm{SO}_{4} 25 \%$ and Glyoxal solution $40 \%$


## Description

## Preparation of the $0.01 \mathrm{~mol} / \mathrm{L}$ lodine solution

The iodine titrant is made out of $0.05 \mathrm{~mol} / \mathrm{l}$ titrant solution which can made from ampuoles.

## Preparation of the Glyoxal solution

A solution of $40 \%$ Glyoxal is adjusted with NaOH to a pH value of 7.0

## Preparation of the $\mathbf{H}_{2} \mathrm{SO}_{4} \mathbf{2 5} \%$

To 75 ml distilled water are carefully added $14 \mathrm{ml} \mathrm{H}_{2} \mathrm{SO}_{4} 96 \%$ (conc.) This mixture is filled up to 100 ml with distilled water.

## Titration

In a 100 or 150 mL beaker are added 50 ml sample and 2 ml of the Glyoxal solution. After 5 minutes 5 ml of the $\mathrm{H}_{2} \mathrm{SO}_{4} 25 \%$ are added. The mixture is directly titrated with the lodine solution to an $\mu \mathrm{A}$ endpoint.

## Methods

## GLP documentation

Titration graph


Method data

Method name:
End date:

Vitamin C with Iodine
17.05.13

Titration duration:
End time:

1 m 12 s
13:32:40

## Titration data

Sample ID:
Start $\mu \mathrm{A}$ :

EP:
$0.846 \mathrm{ml} / 2.5 \mu \mathrm{~A}$

Pattern:
50.000 ml

End $\mu \mathrm{A}$ :

ASC:
$101.31 \mathrm{mg} / \mathrm{l}$

## Calculation formula

ASC:
$(E P-B)^{*} T^{*} M^{*} \mathrm{~F} 1 /\left(\mathrm{V}^{*} \mathrm{~F} 2\right)$
Mol (M):
176.10000
$\begin{array}{ll}\text { Blank value (B): } & 0.0000 \mathrm{ml} \\ \text { Factor 1 (F1): } & 1000.0000 \\ \text { Factor 2 (F2): } & 1.0000\end{array}$
Titre ( T ):
0.03400000 (m)
Pattern (V):
50.000 ml (m)
Statistics:
Off

## sample titration (page 2): <br> Method data overall view

| Method name: | Vitamin C with Iodine | Created at: | 05/17/13 |
| :---: | :---: | :---: | :---: |
| Method type: | Automatic titration | Last modification: | 05/17/13 |
| Measured value: | $\mu \mathrm{A}$ |  |  |
| Titration mode: | d-stop | Documentation: | GLP |
| Linear steps: | 0.020 ml |  |  |
| Measuring speed / drift: | 1 s |  |  |
| Initial waiting time: | 0 s |  |  |
| Titration direction: | Increase |  |  |
| Pretitration: | 0.100 ml | Delay time: | 04 s |
| Endpoint: | $2.5 \mu \mathrm{~A}$ | delta endpoint: | $2.0 \mu \mathrm{~A}$ |
|  |  | Endpoint delay: | 5 s |
| Polarization voltage: | 100 mV |  |  |

Polarization voltage: $\quad 100 \mathrm{mV}$
20.00 \%
4.00 ml

Maximum dosing volume:
Unit values
Unit size:
05 ml
Unit ID:
Reagent:
Batch ID:
Concentration [mol/l]:
Determined at:
Expire date:
Opened/compounded:
Test according ISO 8655:
Last modification:

1296649042
Iod
no Charge
0.03400

10/17/12 22:53:39
12/05/11
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10/17/12 15:53:47

## Hints

If you have any questions concerning the application, you are welcome to contact us.

## Literature

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